

## **Simulation of Diffusion Controlled Intermetallic Formation of Au/Al Interface**

Rui Huang<sup>1</sup>, Yik Yee Tan<sup>2</sup>, Juergen Walter<sup>3</sup>, Heinz Pape<sup>1</sup>, Xuejun Fan<sup>4</sup> and Heinrich Koerner<sup>1</sup>

<sup>1</sup>Infineon Technologies AG, Am Campeon 1-12, 85579 Neubiberg, Germany

<sup>2</sup>Infineon Technologies (Malaysia) Sdn. Bhd., Free Trade Zone, Batu Berendam, 75350 Malacca, Malaysia

<sup>3</sup>Infineon Technologies AG, Wernerwerkstrasse 2, 93049 Regensburg, Germany

<sup>4</sup>Lamar University, Beaumont, Texas 77710, USA

Email: [rui.huang@infineon.com](mailto:rui.huang@infineon.com), Tel: +49 (0)89 234 24929



used as an initial analysis in the subsequent FEM modeling.

Table 2, Material data of Au/Al compounds

<b>Compound</b>	<b>Composition (at. % Au)</b>	<b>Activation energy (eV)</b>	<b>Diffusion coefficient (<math>\mu\text{m}^2/\text{s}</math>)</b>	<b>Density (<math>\text{g}/\text{cm}^3</math>)</b>
Au	84-100			



For intermetallic growth, on the one hand, it is a common belief that during the growth of compounds, the interfacial stresses and stress gradients serves as additional driving force to accelerate

$$V \frac{1}{3V} \frac{dV}{dC}$$



multi-component systems with stoichiometric phases," *Acta Materialia*, vol. 58, pp. 2905-2911, 2010.

- [15] J. Svoboda, E. Gamsjäger, F. Fischer, and E. Kozeschnik, "Modeling of kinetics of diffusive phase transformation in binary systems with multiple stoichiometric phases," *Journal of Phase Equilibria and Diffusion*, vol. 27, pp. 622-628, 2006.

- [16] T. C. Illingworth 19(a)-13(tio)-3(n)a-5(8213(llin( )-809(I)-3O. )-183(I)-3G(ze)-l(ich)s)(tio)-[(Ko5,e )] o14 9.96 Tf1 0 0